GENERAL REMARKS

- The structures are indicated by a double number, the first part of which indicates the chart number in which it occurs and the second part indicates the serial number of the structure, e. g., (1.3) means, structure 3 in Scheme 1.
- 2. The spectra concerning each chapter (2 to 6) are given just before the reference part.
- The ¹H NMR and ¹³CNMR spectra were recorded in CDCl₃ /DMSO-d₆ with TMS as an internal reference; on Bruker AC-200 or MSL-300 (200 MHz or 300 MHz for ¹H NMR and 50 MHz or 75 MHz for ¹³C NMR) spectrometer. The chemical shifts are expressed in δ units.
- 4. The IR spectra were recorded in CHCl₃, nujol or as a KBr pellet on Perkin- Elmer- 783 spectrophotometer and the values are expressed in cm⁻¹.
- The mass spectra were recorded on Phenegan Matt SSQ 7000 Spectrometer.
- 6. The melting points (m. p.) are uncorrected.
- 7. The ether extracts were dried with anhydrous sodium sulphate, unless otherwise mentioned.
- 8. The abbreviations used in the literature and charts are as given below:

ABBREVATIONS:

Ac	Acetyl
AcCN	Acetonitrile
CAN	Ceric Ammonium Nitrate
Cat.	Catalyst

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DCM	Methylene chloride
DDQ	Dichloro dicyano-p-Benzoquinone
DMF	Dimethyl Formamide
DMSO	Dimethyl Sulfoxide
g	Grams
h.	Hour
HMDS	Hexamethylenedisilazane
min.	Minutes
mol	Mole
mmol	Milli mole
SA	Sulfamic acid
MW	Microwave
NBS	N-Bromosuccinimide
LiBF ₄	Lithium tetrafluoroborate
PEG	Polyethylene glycol
РТС	Phase transfer catalyst
Ру	Pyridine
rt	Room temperature
SSA	Silica sulfuric acid
TBDMS	tert-butyl dimethyl silyl
TCICA	Trichloroisocyanuric acid
TEA	Triethylamine
TLC	Thin layer chromatography
THF	Tetrahydrofuran
THP	Tetrahydropyran

TMSCN Trimethyl silyl cyanide

- MCRs Multicomponent reactions
- PVP-HCl Polyvinyl pyridine hydrochloride